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(54) Title: METHOD OF OBTAINING (22R) DIASTEREOISOMER OF BUDESONIDE

(57) Abstract

By the method according to the invention condensation of 11β , 16α , 17α , 21-tetrahydroxy-1,4-pregnadiene-3,20-dione 21-acetate with n-butyric aldehyde is carried out, in the known way, in the medium of hydrofluoric acid of concentration of 70-80 %. Theisolated crude condensation product is crystallized from ethanol and obtained 21-acetate of budesonide (22R) of at least 95 % content is hydrolyzed, and the product thus obtained is crystallized from ethyl acetate.

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Method of obtaining (22R) diastereoisomer of budesonide

The subject of the invention is a method of obtaining diastereoisomer of (22R) budesonide. Budesonide, i.e. 5 $(22R,S)-16\alpha,17\alpha$ -butylidenedioxy-11 $\beta,21$ -dihydroxy-1,4-pregnadiene-3,20-dione is a mixture of diastereoisomers (22R) and (22S) differing in the position of an acetal (fig.1 and 2). Both compounds are chain glucocorticoids mixture (1:1)applied in in а forms: antiasthmatic aerosol 10 pharmaceutical antiallergic ointment. (22R) diastereoisomer is 2-3 times more active pharmacologically than (22S) diastereoisomer [Brattsand R.; Eur.J.Res.Dis.63, suppl. 122,62-73 (1982).

In pharmacotherapy a tendency is presently observed to apply optically active compounds and not racemic mixtures. In the case of budesonide both isomeric forms are pharmacologically active but their metabolism is different [Andersson P. et al.; Xenobiotica 17,35-44 20 (1987)]. Testing of action of a drug being a pure chemical individual is considerably easier than that of a mixture.

A known method of obtaining budesonide, that is a mixture of diastereoisomers (22R) and (22S) consists in 25 condensation of 11β,16α,17α,21-tetrahydroxy-1,4-pregnadien-3,20-dione with n-butyric aldehyde in the presence of strong inorganic acids, e.g. perchloric acid, in organic solvents [FRG patent 2323216(1973)].

Those skilled in the art know also a method of 30 obtaining a mixture of diastereoisomers (22R) and (22S), in which isomer (22R) is in majority, even up to 90%. The said method consists in condensation, discussed above, conducted in the presence of hydrofluoric acid of concentration of 48-70% or concentrated hydrochloric acid 35 [Eur.pat.appln.164636 (28.05.85)].

According to the known method separation of a mixture of diastereoisomers (22R) and (22S) is carried out on a

column packed with Sephadex LH20 [Thalen A., Nylander B.; 19,247-266(1982)]. The said method requires the use of a very big amount of solvents and long separation time, and for these reasons it is technologically inconvenient, 5 expensive and time-consuming.

Unexpectedly it has appeared out that the product obtained after condensation in the form of 21-acetate containing at least 80% of diastereoisomer (22R), after crystallization from ethanol, and then hydrolyzed and yields pure acetate ethyl from 10 crystallized admixture of with an (22R)diastereoisomer diastereoisomer (22S) of 1% at the very most. The content of diastereoisomers is determined by the HPLC method. From post-crystallization filtrates containing a mixture 15 of diastereoisomers of a composition of 8:2 or 7:3 one may separate budesonide 21-acetate, crystallize it again and after hydrolysis obtain pure diastereoisomer (22R).

By the method according to the invention condensation of 11\$\beta\$, 16\$\alpha\$, 16\$\alpha\$, 17\$\alpha\$, 21-tetrahydroxy-1, 4-pregnadiene-3, 20-dione 20 21-acetate with n-butyric aldehyde is carried out, in the known way, in the medium of hydrofluoric acid of concentration of 70-80\$\%2\$. The isolated crude condensation product is crystallized from ethanol and obtained 21-acetate of budesonide (22\$\Reg\$) of at least 95\$\%2\$ content is 25 hydrolyzed, and the product thus obtained is crystallized from ethyl acetate to obtain (22\$\Reg\$) diastereoisomer of budesonide of at least 99\$\%2\$ content.

An advantageous effect of the invention is that (22R) diastereoisomer of budesonide having the content of at 30 least 99% is obtained at a yield of about 60%. The product obtained by the method according to the invention is an active substance of antiasthmatic aerosol and its content in a dose of 158 µg is equivalent to 200 µg of budesonide of a mixture (1:1) of diastereoisomers (22R 35 and 22S). The method according to the invention is characterized by simplicity of procedure, enables saving organic solvents and is less labour-consuming.

The after-mentioned example illustrates the invention without limiting its scope.

Example.

To a mixture of 7 ml of 75% hydrofluoric acid and 5 0.52 ml of n-butyric aldehyde, cooled to 0°C, 3.5 g of 11β , 16α , 17α , 21-tetrahydroxy-1, 4-pregnadiene-3, 20-dione 21-acetate is added in and stirred for 3 hours at maintenance of the temperature of 0°C. The obtained dark-red solution is poured into 50 ml of water with ice, 10 neutralized by concentrated ammonia and extracted by chloroform. From the extract chloroform is distilled off the residue is and diminished pressure, crystallized from ethanol. 2.2 g of 21-acetate of (22R) diastereoisomer of budesonide of $[\alpha]_D^{22}+106^{\circ}(c=1, CH_2Cl_2)$ 15 is obtained, the content of 21-acetate of (22S) diastereoisomer of budesonide is 5%. The product is suspended in 40 ml of methanol, cooled to 0°C, 2.5 ml of 10% aquas solution of potassium carbonate is added thereto, and is stirred under nitrogen for 1 and 1/2 20 hours, at maintenance of the temperature of 0° C, after this it is neutralized with acetic acid, methanol is evaporated and (22R) diastereoisomer of budesonide is isolated either by filtering off or by extraction with chloroform. The separated product is crystallized from 25 ethyl acetate and 1.7 g of (22R) diastereoisomer of budesonide is obtained, of melting point of 245-250°C (decomp.) $[\alpha]_D^{22} + 117.5^{\circ}C$ (c=1, CH_2Cl_2); a_{1cm}^{12} 350 at 242 nm. The content of (22S) diastereoisomer of budesonide determined by the the HPLC method is 1%.

Patent claim

Method of obtaining of (22R) diastereoisomer of budesonide, consisting in a condensation reaction of 11β,16α,17α,21-tetra-hydroxy-1,4-pregnadiene-3,20-dione 21-acetate with n-butyric aldehyde conducted in the medium of hydrofluoric acid of concentration of 70-80%, characterized in that the isolated crude product of condensation is crystallized from ethanol and obtained 10 21-acetate of budesonide (22R) of at least 95% content is hydrolyzed, and the product thus obtained is crystallized from ethyl acetate.

Fig. 1

Fig. 2

International Application No

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	cited i	n the application		
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Α	US,A,3 996 359 (AB BOFORS) 7 December 1976			1
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ANNEX TO THE INTERNATIONAL SEARCH REPORT ON INTERNATIONAL PATENT APPLICATION NO. PL 9100016 54730

This annex lists the patent family members relating to the patent documents cited in the above-mentioned international search report.

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Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP-A-0164636	18-12-85	JP-C- 1588637 JP-B- 2013680 JP-A- 61040299 US-A- 4835145 US-A- 4695625	04-04-90 9 26-02-86 5 30-05-89
US-A-3996359	07-12-76	SE-B- 378110 AU-A- 5525273 BE-A- 799728 CA-A- 1003402 DE-A, C 2323216 FR-A, B 2185406 GB-A- 1428416 JP-C- 975395 JP-A- 49041379 JP-B- 54007794 NL-A- 7306979 US-A- 3928326	3 07-11-74 8 17-09-73 2 11-01-77 6 29-11-73 6 04-01-74 6 17-03-76 5 19-10-79 9 18-04-74 4 10-04-79 9 21-11-73 6 23-12-75